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achieves Suzuki cross-couplings of an array of nitrogencontaining boronic acids and aryl halides [Eq. (1); dba = dibenzylideneacetone, Cy = cyclohexyl].

heteroaryl-B(OH)₂ X-Ar
$$(1.1 \text{ equiv}) \quad Ar = \underset{\text{heteroaryl}}{\text{Ar}} = \underset{\text{heteroaryl}}{\text{A$$

Since 1998, we and others have established that trialkylphosphines, including P(tBu)₃ and PCy₃, serve as very effective ligands for a variety of palladium-catalyzed coupling processes. [1,5] In view of the low cost of PCy3 relative to many of the other ligands that provide highly active cross-coupling catalysts (e.g., $P(tBu)_3$, aryldialkylphosphines, and carbenes), we decided to examine the possibility of developing a versatile PCy₃-based catalyst for Suzuki reactions of nitrogen heterocycles.

In an earlier study, we employed Pd/PCy₃/KF/THF as a catalyst system for Suzuki cross-couplings of aryl halides.^[6,7] Unfortunately, when we applied these conditions to the coupling of 3-pyridineboronic acid with 4-n-butylchlorobenzene, we obtained essentially none of the desired biaryl [Eq. (2)]. After considerable exploration of the reaction parameters, we determined that Pd/PCy₃/K₃PO₄/dioxane/ H₂O achieves the desired cross-coupling in excellent yield [Eq. (2)]. [8,9]

$$\bigcap_{N} B(OH)_2 \bigcap_{CI} nBu$$
 (2)

1.5% [Pd₂(dba)₃], 4.2% PCy₃, 3.3 equiv KF, THF, 60 °C 1.0% [Pd₂(dba)₃], 2.4% PCy₃, 1.7 equiv K₃PO₄, dioxane/H₂O, 100 °C

We were pleased to find that this standard protocol can be applied directly to Suzuki reactions of an array of nitrogen heterocycles. As illustrated in Table 1, a variety of heteroarylboronic acids can even be coupled with challenging aryl halides. Thus, 3-pyridineboronic acid reacts with a range of chlorobenzenes, including ortho-substituted and electronically deactivated substrates, in good yield (Table 1, entries 1– 4). As might be anticipated, the corresponding aryl bromides and iodides also couple under these conditions (Table 1, entries 5 and 6). In addition, substituted 3-pyridineboronic acids are suitable cross-coupling partners (Table 1, entries 7– 9). The method is effective not only for 3-pyridineboronic acids, but also for an array of other heteroarylboronic acids (e.g., 4-pyridine, 5-pyrimidine, and indole; Table 1, entries 10-13).

Although substrates that bear an indole NH group undergo Suzuki cross-coupling smoothly with this procedure (Table 1, entry 13), the catalyst appears to be sensitive to pyrazole NH groups (Table 1, entries 14 and 16). However, simple protection of the nitrogen atom leads to coupling with useful efficiency (Table 1, entries 15 and 17).[10]

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A Versatile Method for Suzuki Cross-Coupling Reactions of Nitrogen Heterocycles**

Noriaki Kudo, Mauro Perseghini, and Gregory C. Fu*

The Suzuki cross-coupling reaction is a powerful method for carbon-carbon bond formation that has been applied in a variety of settings, ranging from natural-products synthesis to materials chemistry, including large-scale production. [1,2] Among the attractive features of the Suzuki reaction are the wide availability, stability to air and moisture, and low toxicity of boronic acids, as well as the facile removal of the boron-containing side products of the coupling process.

In recent years, tremendous progress has been described in the discovery of more active catalysts for the Suzuki reaction. Arguably the most important remaining challenge is the development of a universal method for cross-coupling substrates that include nitrogen heterocycles.[3] The presence of such groups, which are particularly pervasive in medicinal chemistry, [4] can lead to low reactivity in coupling reactions. Herein, we describe an unusually versatile catalyst that

[*] Dr. N. Kudo, Dr. M. Perseghini, Prof. Dr. G. C. Fu Department of Chemistry Massachusetts Institute of Technology Cambridge, MA 02139 (USA) Fax: (+1) 617-324-3611 E-mail: gcf@mit.edu

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Table 1: Suzuki cross-couplings of heteroarylboronic acids with unactivated aryl halides [for the reaction conditions, see Eq. (1)].

Entry	Heteroarylboronic acid	Aryl halide	Yield [%] ^[a]
1	B(OH) ₂	CI	92
2	B(OH) ₂	n-Bu	92
3	B(OH) ₂	Me	95
4	B(OH) ₂	MeO	88
5	B(OH) ₂	Br	90
6	B(OH) ₂		91
7	B(OH) ₂ OMe	CI	83
8	MeO N B(OH) ₂	CI	87
9	CI N B(OH) ₂	Br	85 ^[b]
10	B(OH) ₂	CI	92
11	N B(OH) ₂	CI	72
12	B(OH) ₂	CI	85
13	(HO) ₂ B	CI	78
14	B(OH) ₂	CI	8
15	B(OH) ₂ N N Bn	CI	64
16	NN NB(OH) ₂	Br	21
17	N _N B(OH) ₂	Br	92

[a] Yield of isolated product (average of two experiments). [b] The product that results from the cross-coupling of the chloride was isolated in 10% yield.

This catalyst system is also effective for Suzuki cross-couplings in which both of the reaction partners are nitrogen heterocycles, thus providing the desired products in generally good yield (Table 2). Consistent with the example illustrated in entry 13 of Table 1, the method tolerates indole NH groups

Table 2: Suzuki cross-couplings of heteroarylboronic acids with heteroaryl chlorides and bromides [for the reaction conditions, see Eq. (1)].

Entry	Heteroarylboronic acid	Heteroaryl halide	Yield [%] ^[a]
1	B(OH) ₂	CIN	97
2	B(OH) ₂	CI N H	93
3	B(OH) ₂	N · HCI	98
4	B(OH) ₂	CINCI	98 ^[b]
5	B(OH) ₂	Br	75
6	B(OH) ₂	OH CI N	87
7	B(OH) ₂	OBn CI N	88
8	B(OH) ₂	H_2N	97
9	MeO N B(OH) ₂	CI NH2	95
10	B(OH) ₂	CIN	97
11	N B(OH) ₂	CIN	77
12	(HO) ₂ B	CIN	89
13	Boc	CIN	69 ^[c]
14	Boc B(OH) ₂	Br	30 ^[c]
15	B(OH) ₂	H N N Br	< 2
16	B(OH) ₂	Bn N N Br	73

[a] Yield of isolated product (average of two experiments). [b] Double Suzuki cross-coupling (2.2 equiv of the boronic acid, 2.0% [Pd₂(dba)₃], 4.8% PCy₃, and 3.4 equiv of K_3 PO₄ were used). [c] The Boc protecting group was removed during the Suzuki reaction.

(Table 2, entries 2 and 12). In addition, pyridines that bear hydroxy (Table 2, entry 6) and NH₂ (Table 2, entries 8 and 9) substituents are suitable substrates. *tert*-Butyloxycarbonyl (Boc) protecting groups at pyrrole and indole nitrogen atoms are cleaved under the cross-coupling conditions

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(entries 13 and 14, respectively). [11] Protection of the pyrazole NH group is necessary for the coupling of a 4-bromopyrazole with 3-pyridineboronic acid (entry 15 vs. 16; see also Table 1, entries 14–17).

In preliminary experiments, we determined that this Pd/PCy₃/K₃PO₄/dioxane/H₂O-based method is effective not only for Suzuki cross-couplings of heteroarylboronic acids, but also for boronate esters and trifluoroborates [Eq. (3)]. Finally, we

established that the procedure can be conducted on a multigram scale (18 mmol of 4-n-butylchlorobenzenechloride + 20 mmol of 3-pyridineboronic acid, yield = 3.4 g (88%) of product).

To the best of our knowledge, this is the most wideranging study that has been described for Suzuki reactions of nitrogen-containing cross-coupling partners. Attractive features of this method include its versatility (a single procedure was employed for all of the examples, including boronate esters and trifluoroborates), its compatibility with a variety of unprotected functionalities (e.g., NH₂- and OH-substituted pyridines and unprotected indoles), its convenience (commercially available components), and its efficiency even with inexpensive, unactivated aryl chlorides. We anticipate that this catalyst system will find application in academia and, in particular, in industry.

Experimental Section

General procedure: The heteroarylboronic acid (1.10 mmol), $[Pd_2-(dba)_3]$ (9.2 mg, 0.010 mmol), and PCy_3 (6.7 mg, 0.024 mmol) were added to a 25-mL Schlenk flask equipped with a stir bar in air. The flask was evacuated and refilled with argon five times. Dioxane (2.67 mL), the (hetero)aryl halide (1.00 mmol; if the halide was a solid, it was added prior to the evacuation/refill cycle), and aqueous K_3PO_4 (1.27 m, 1.33 mL, 1.70 mmol) were added by syringe. The Schlenk flask was sealed and heated in an oil bath at 100°C for 18 h with vigorous stirring. The mixture was then filtered through a pad of silica gel (washing with EtOAc), the filtrate concentrated under reduced pressure, and the aqueous residue extracted three times with EtOAc. The combined extracts were dried over anhydrous MgSO₄, filtered, and concentrated. The residue was then purified by column chromatography on silica gel.

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